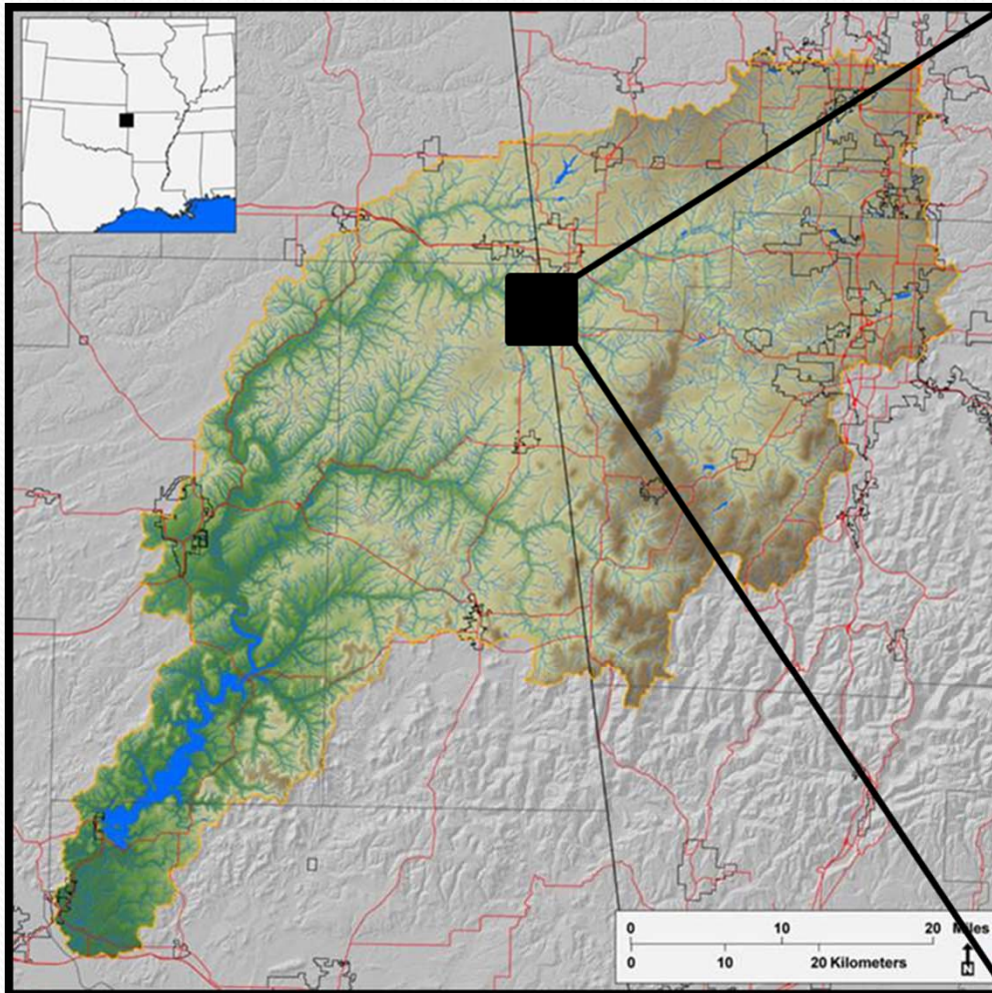


Lake Frances

How Much Alum is Enough?

Jason Corral and Brian E. Haggard
Department of Biological and Agricultural Engineering
University of Arkansas, Fayetteville, AR
Email: jcorral@uark.edu

Study Site





Background

- Oklahoma Scenic Rivers like the Illinois River have a total phosphorus (TP) criterion of 0.037 mg L^{-1}
- Previous study showed high potential for phosphorus (P) release from sediment
 - Average P flux of 3.6 and $15.5 \text{ mg-P m}^{-2} \text{ d}^{-1}$ under aerobic and anaerobic conditions, respectively (Haggard and Soerens, 2006)
- These release rates are high relative to that measured in regional reservoirs:
 - Lake Eucha: 1.03 and $4.40 \text{ mg-P m}^{-2} \text{ d}^{-1}$ under aerobic and anaerobic conditions, respectively (Haggard et al, 2005)
 - Lake Wister: 0.94 and $2.41 \text{ mg-P m}^{-2} \text{ d}^{-1}$ under aerobic and anaerobic conditions, respectively (Haggard and Scott, 2011)
 - Beaver Lake: 0.15 and $1.77 \text{ mg-p m}^{-2} \text{ d}^{-1}$ under aerobic and anaerobic conditions, respectively (in the Transition Zone) (Sen et al, 2007)

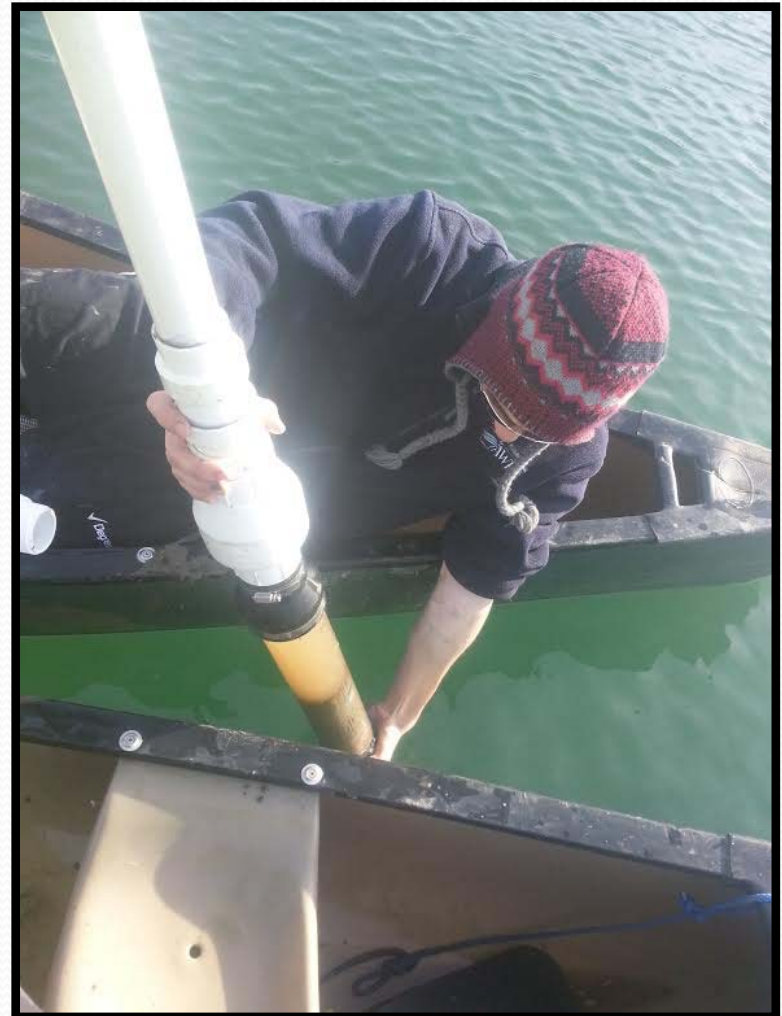


What Needs to be Answered?

- Is P release still occurring?
- Is the P contribution significant?
- Can we do something about it?
 - Conventional treatments include:
 - Chemical: for example, alum ($\text{Al}_2(\text{SO}_4)_3 \cdot 18 \text{H}_2\text{O}$) application
 - Physical: for example, sediment dredging
- If we want to do something about it, what exactly should we do?

Field Methods

- Collect Cores
 - Want approximately 20 cm of sediment
 - Adjust to 1 L of overlying water at lab
- Stopper both ends
- Cores collected near dam
 - ~6 m deep
 - ~0.6 m sediment depth



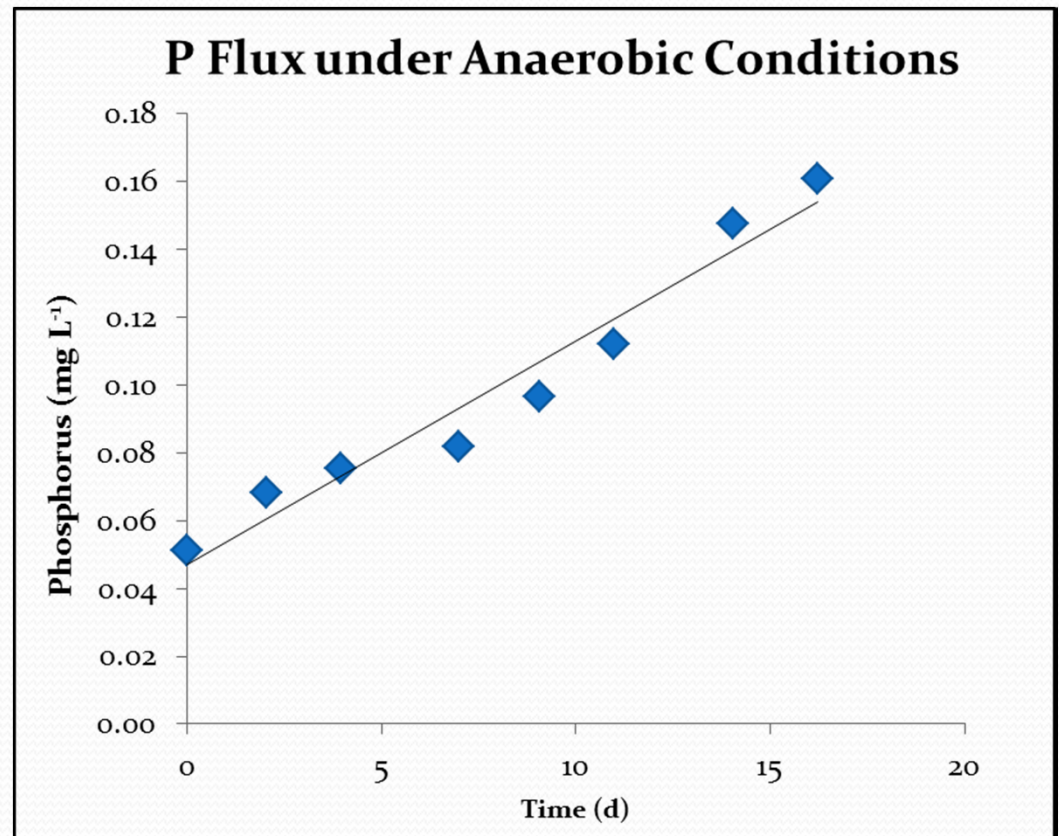
Lab Methods

- Incubate under aerobic and anaerobic conditions
 - Anaerobic conditions create redox potential necessary for accelerated P release
- Monitor P concentration for 15 days
- Determine P flux ($\text{mg-P m}^{-2} \text{d}^{-1}$)
- Treat with various amounts of alum
- Continue incubation for another 15 days
- Monitor P concentration
- Determine P flux with alum treatment



Determining Flux

- Example plot of P vs time
- Calculate slope, divide by surface area for flux
 - The surface area of the sediment-water interface





Treatment

What Questions Need to be Answered?

- Alum addition
 - Treatment dose determined by P flux.
 - What alum dose is the best treatment option?
 - Test a range of doses
 - Look at initial P sequestration
 - Look at long term P flux mitigation
- Important notes
 - 2 mols Al per mol alum
 - 18 mols H₂O per mol alum

Adding Alum

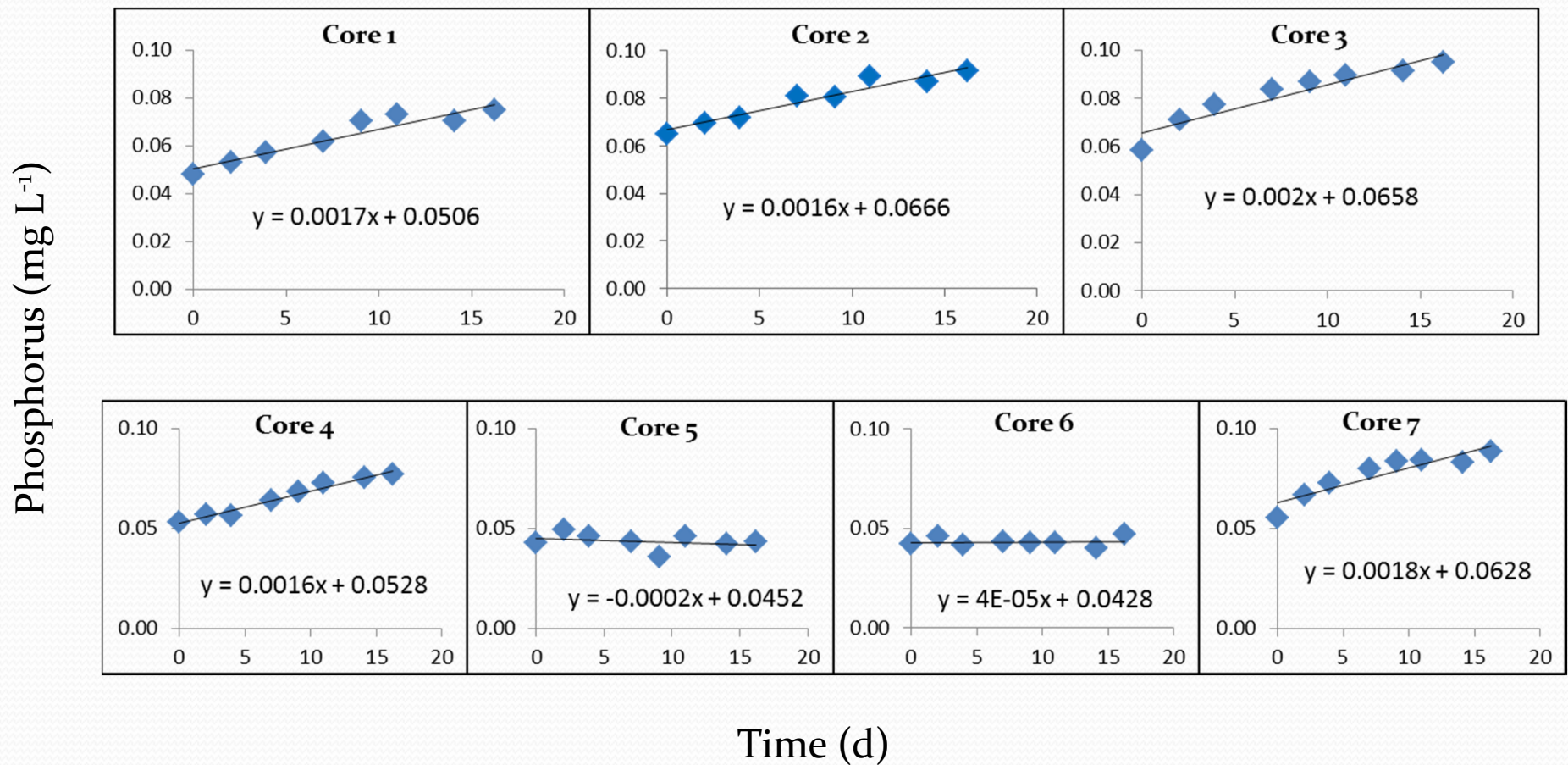
- When alum is added to water, it forms a floc
- As the floc settles, it removes dissolved P from the water column
- This floc forms a sediment “cap” which can continue to capture dissolved P as it is released

Treatments Tested	
Areal Basis (g-alum m ⁻²)	Concentration (mg-Al L ⁻¹)
0	0
6.7	2.5
13.5	5
26.9	10
53.8	20
108	40
269	100



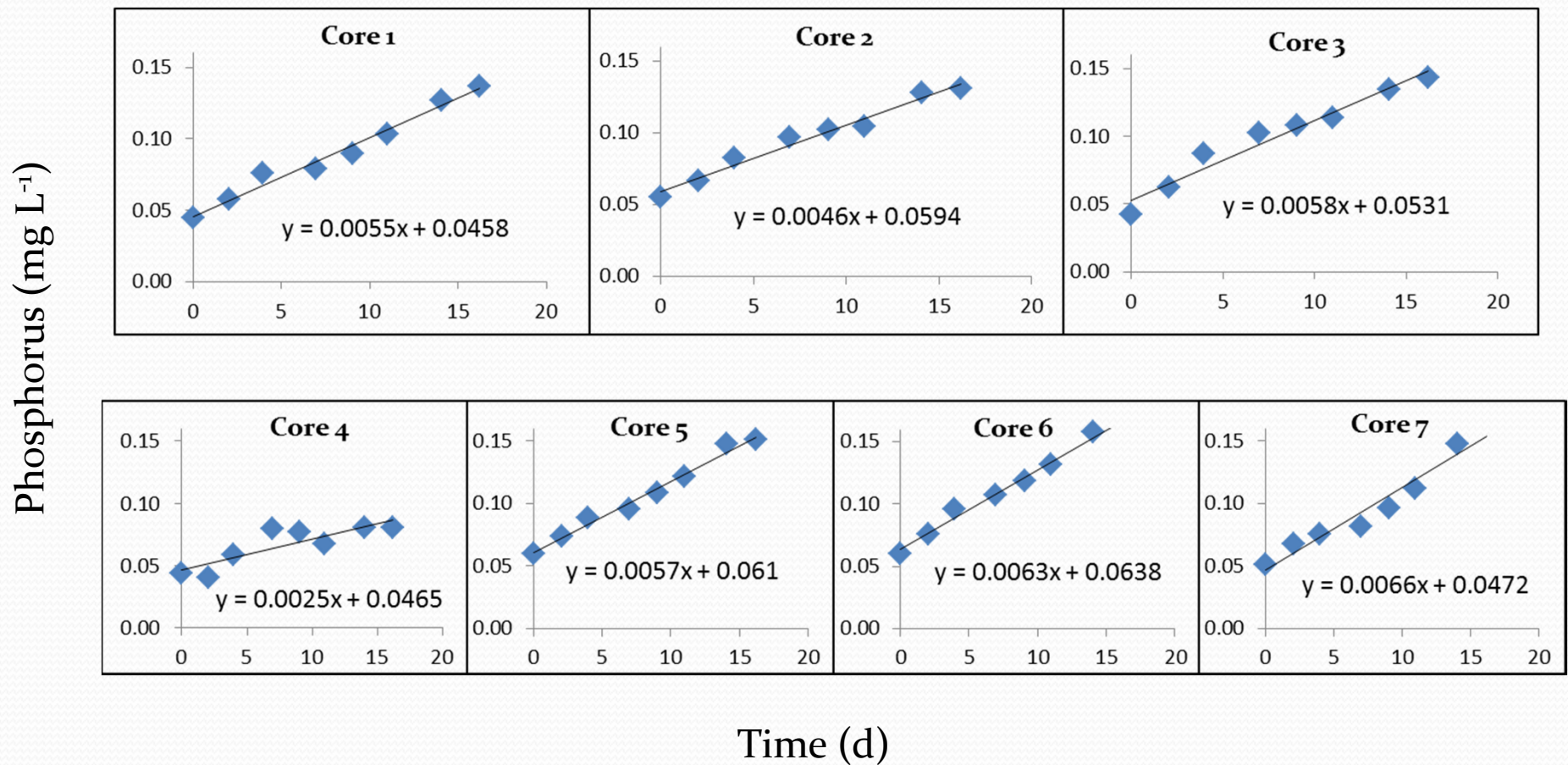
Results

Aerobic Flux Before Alum

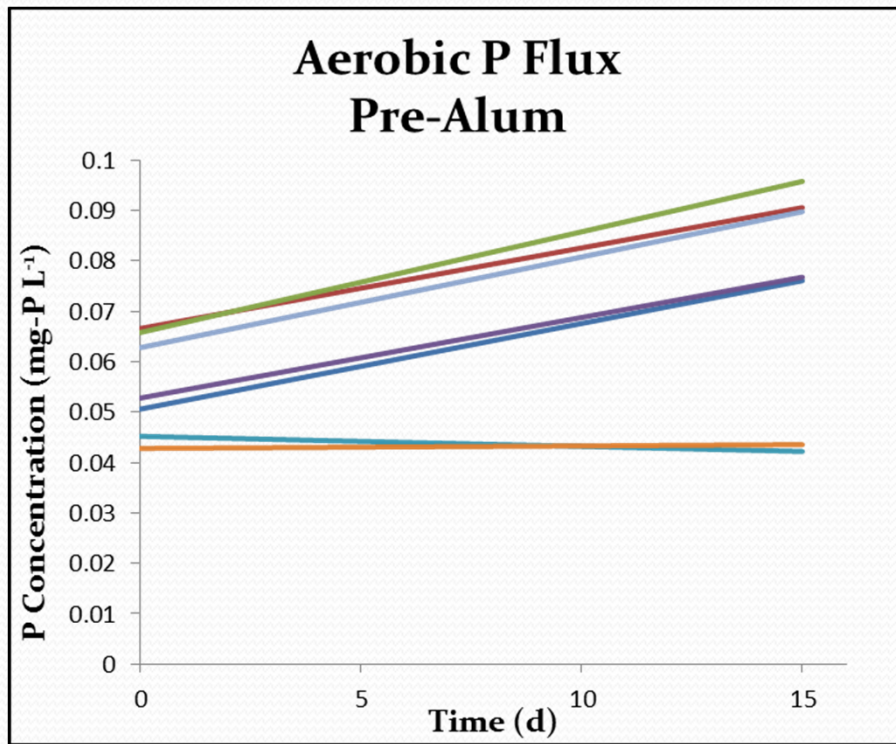


Results

Anaerobic Flux Before Alum

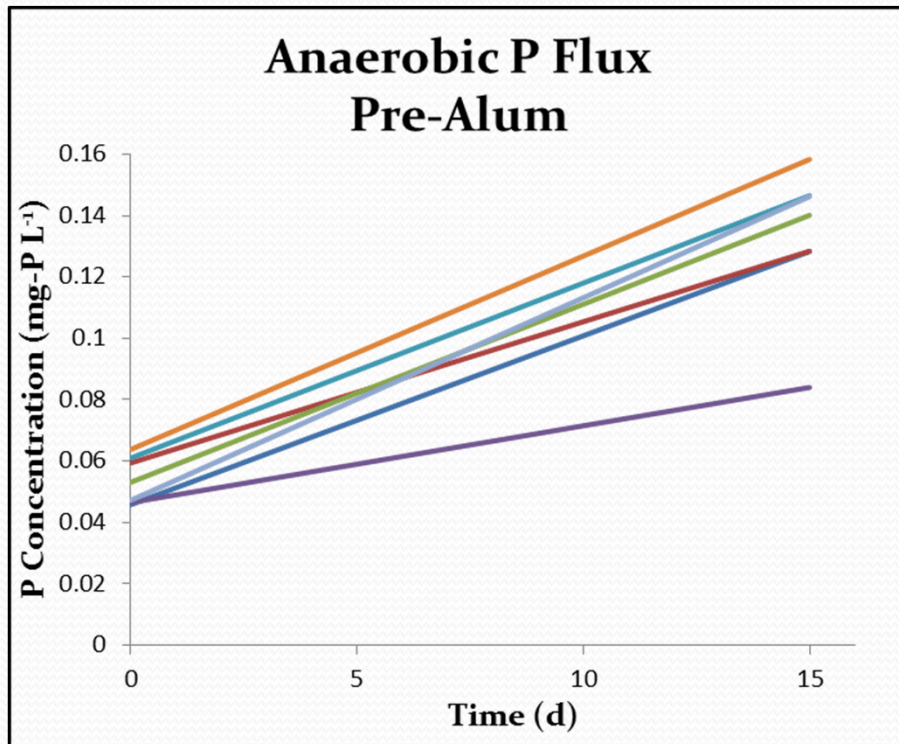


So, is P Release Still Occurring?



Core	P Flux (mg-P m ⁻² d ⁻¹)
1	0.37
2	0.35
3	0.44
4	0.35
5	-0.04
6	0.01
7	0.39
Average	0.27

So, is P Release Still Occurring?



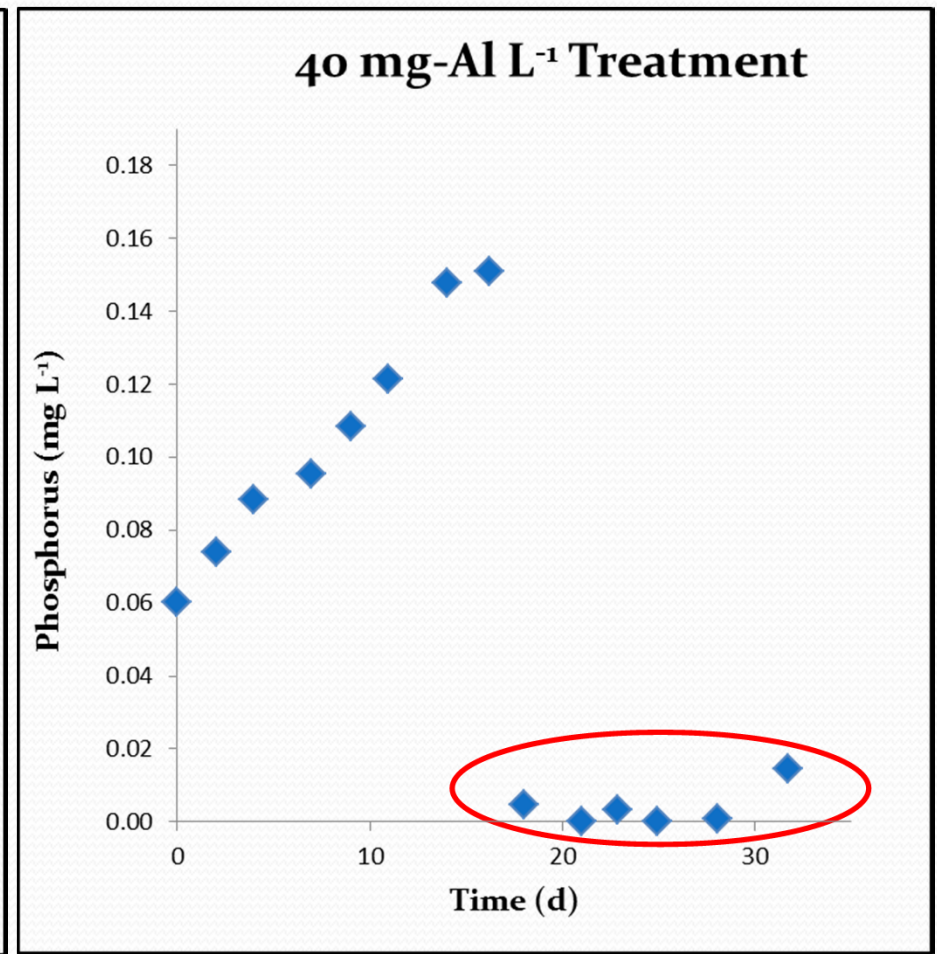
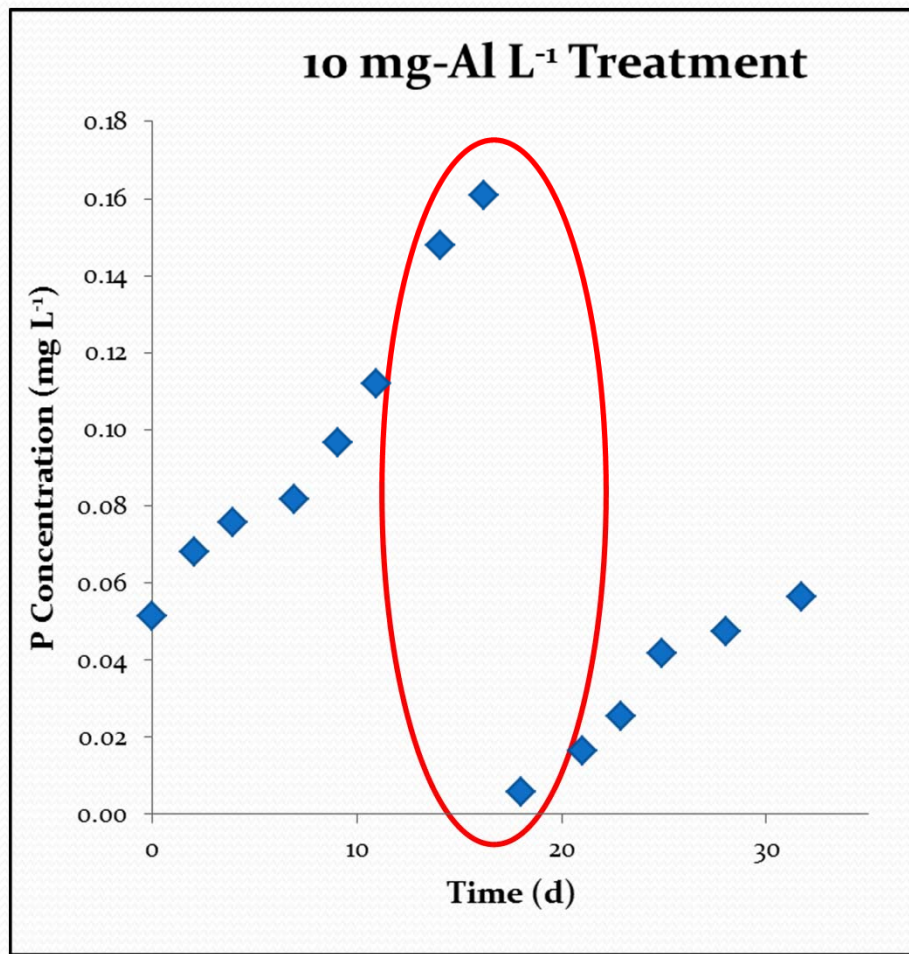
Core	P Flux (mg-P m ⁻² d ⁻¹)
1	1.20
2	1.00
3	1.27
4	0.56
5	1.24
6	1.37
7	1.44
Average	1.15



Is This P Release Significant?

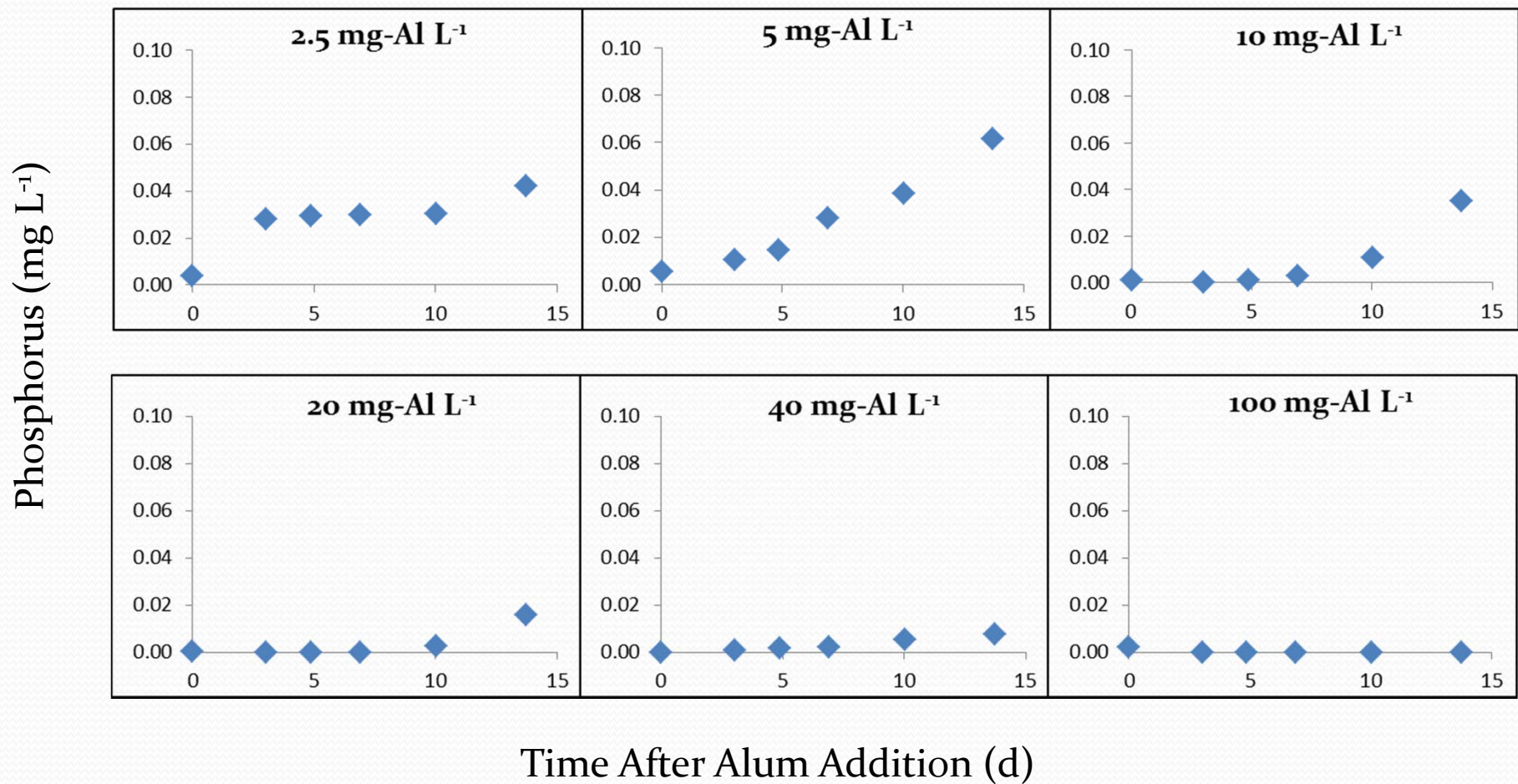
- Average P flux under aerobic ($0.27 \text{ mg m}^{-2} \text{ d}^{-1}$) and anaerobic ($1.15 \text{ mg m}^{-2} \text{ d}^{-1}$) conditions are less now
 - Previously $3.6 \text{ mg m}^{-2} \text{ d}^{-1}$ under aerobic conditions
 - Previously $15.5 \text{ mg m}^{-2} \text{ d}^{-1}$ under anaerobic conditions
 - Haggard and Soerens (2006)
- But, even under aerobic conditions, P release occurs until equilibrium interactions take effect around 0.10 mg-P L^{-1}
- If water in the Illinois River enters Lake Frances at $0.037 \text{ mg-P L}^{-1}$, P release will occur!

Alum Addition at a Glance



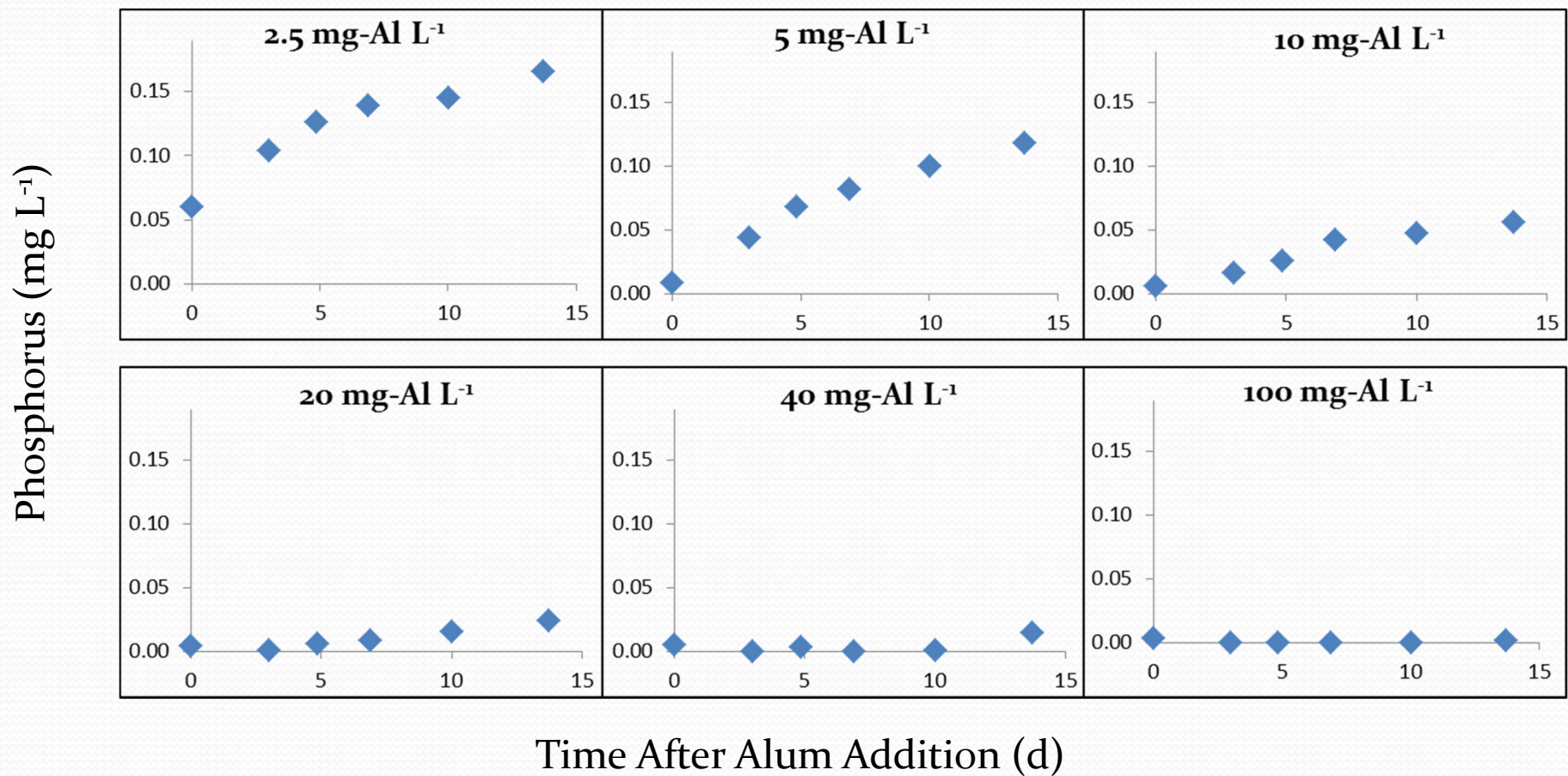
Results

Aerobic Flux with Alum

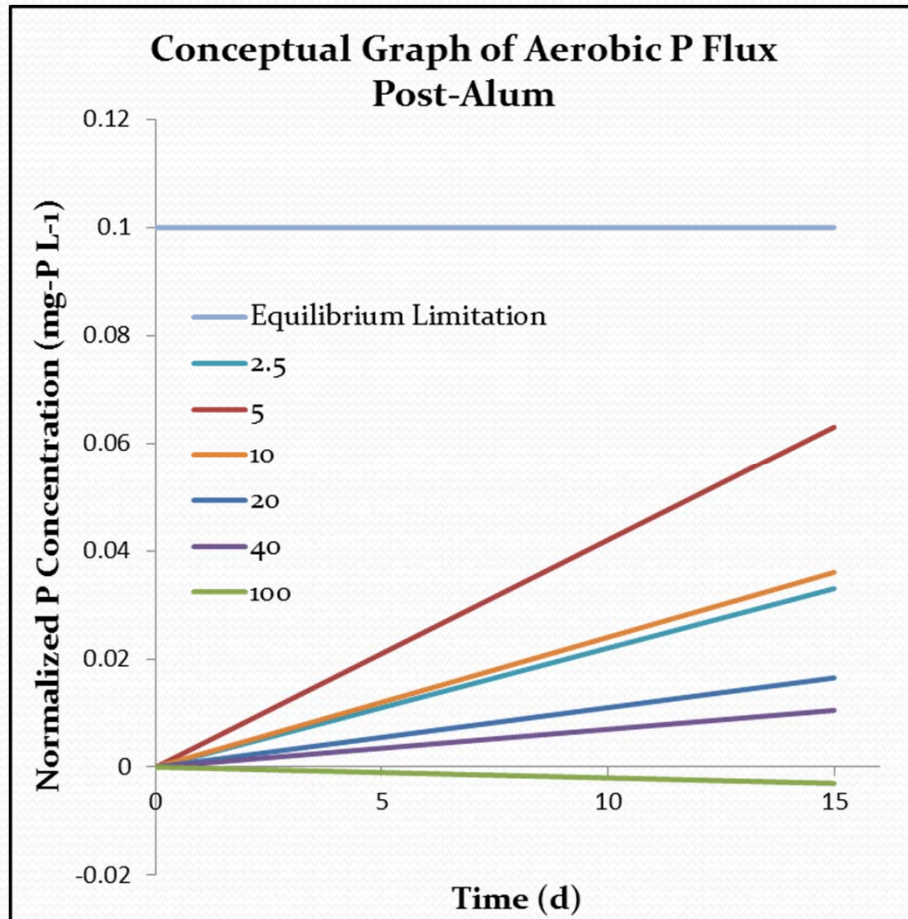


Results

Anaerobic Flux with Alum

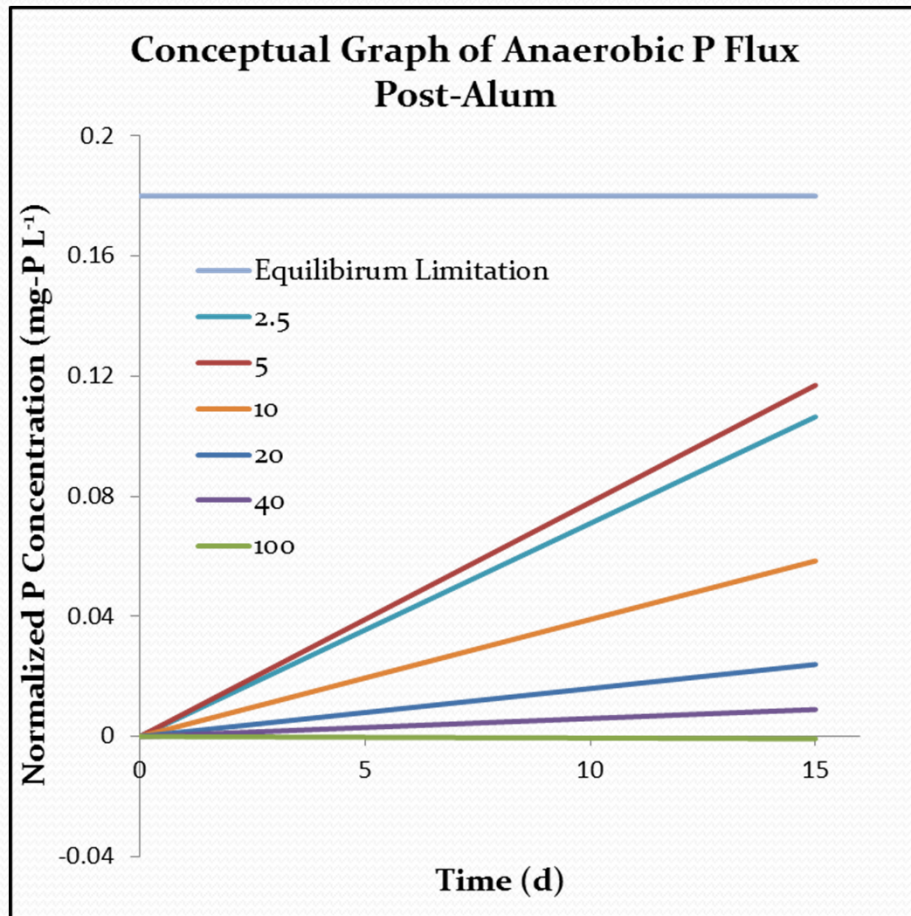


So, Can We Do Something About It?



Treatment (mg-Al L ⁻¹)	P Flux (mg-P m ⁻² d ⁻¹)
2.5	0.48
5	0.92
10	0.52
20	0.24
40	0.15
100	-0.04

So, Can We Do Something About It?



Treatment (mg-Al L^{-1})	P Flux ($\text{mg-P m}^{-2} \text{d}^{-1}$)
2.5	1.55
5	1.70
10	0.85
20	0.35
40	0.13
100	-0.01

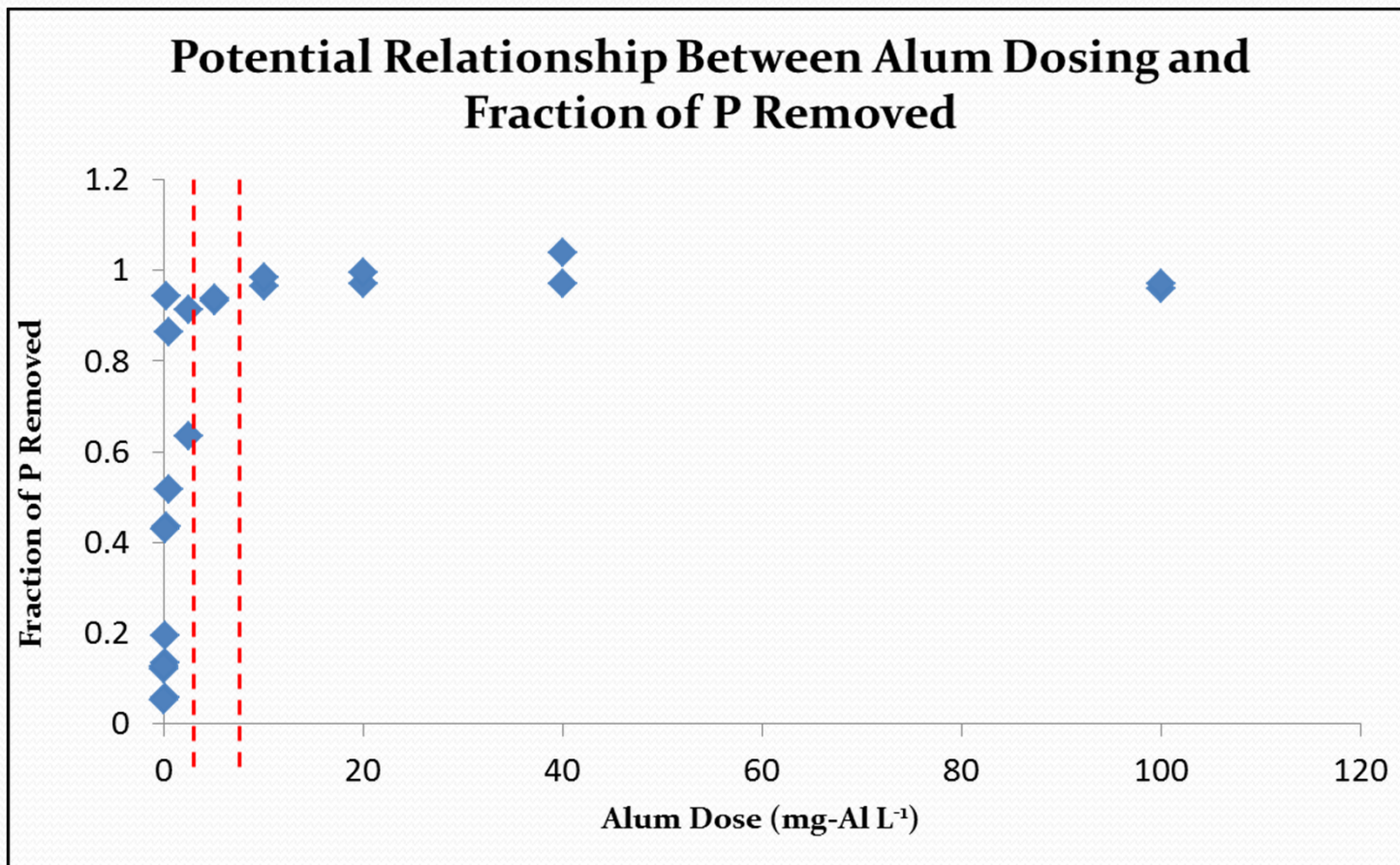


What to do with Results?

- There are 2 measures of success, depending on your desired effect:
 - How much dissolved P was removed from the water column?
 - Was P flux mitigated?

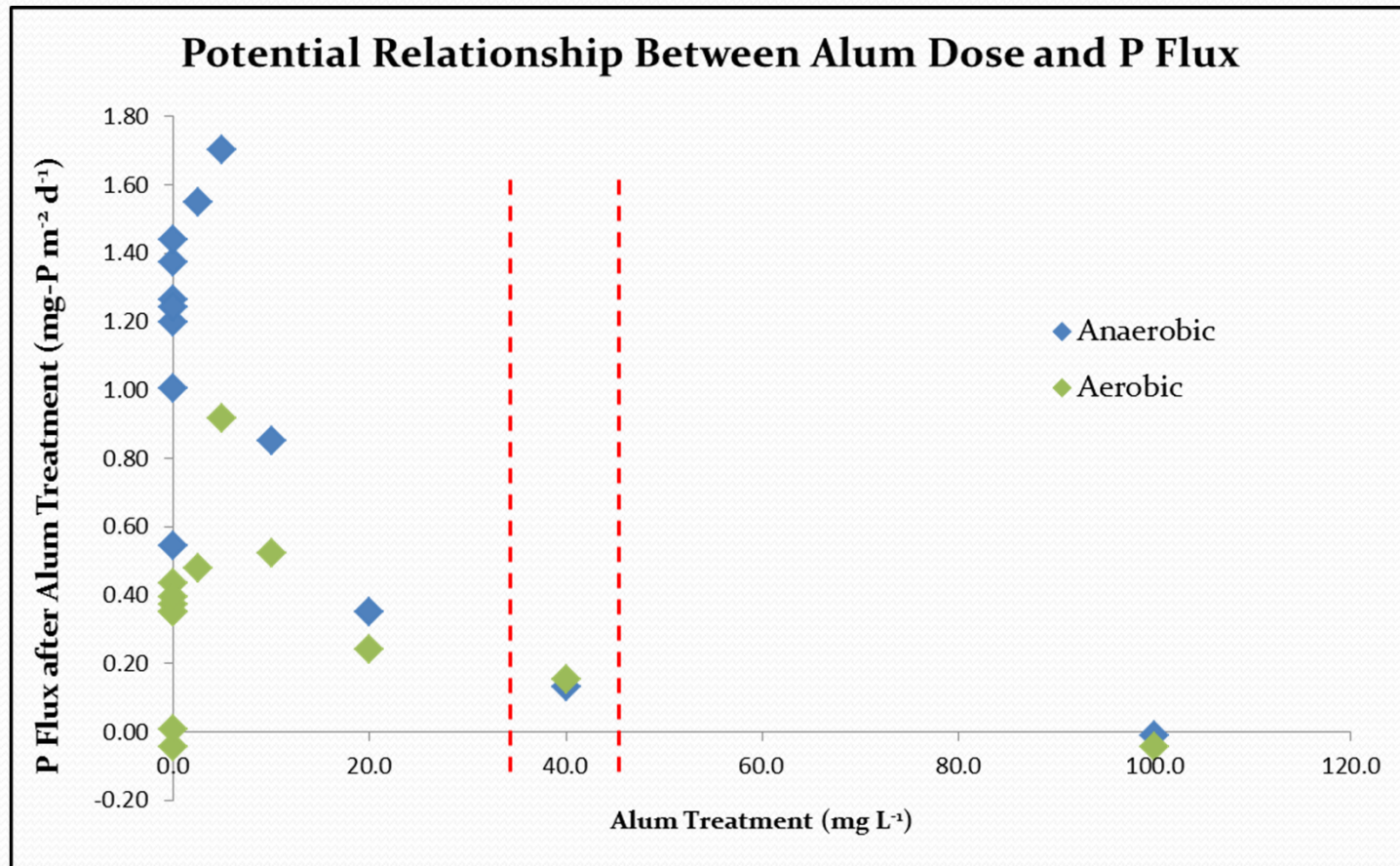
Results

P Removal by Alum



Results

P Flux Mitigation by Alum





So, What Exactly Should We Do?

- For a one-time treatment of P in the water column of Lake Frances: dose Alum at $2.5 - 5 \text{ mg-Al L}^{-1}$
- For long term sediment treatment at Lake Frances: dose Alum at approximately $108 \text{ g-alum m}^{-2}$ (40 mg-Al L^{-1})



Did We Answer All of Our Questions?

- Is P release still occurring?
 - Yes
- Is the P contribution significant?
 - Yes
- Can we do something about it?
 - Yes
- If we want to do something about it, what exactly should we do?
 - 2.5 – 5 mg-Al L⁻¹ for treatment of the water column
 - Around 108 g-alum m⁻² (40 mg-Al L⁻¹) for long term P flux mitigation



Acknowledgements

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Questions?

